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Thalidomide metabolites and analogs. Part 2: Cyclic derivatives of 2-N-phthalimido-2S,3S (3-hydroxy) ornithine

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Abstract

E-(5-N-phthaloyl)-2-pentenoic acid benzyl ester was dihydroxylated by a Sharpless AD-mix protocol followed by mononosylation of the resultant 2R,3S diol. Azide displacement of the mononosylate followed by protection with the TBDMS group gave the ε -N-phthaloyl-substituted- β -tert-butyldimethylsilyloxy- α azidocarboxylic acid benzylester. Hydrazinolysis resulted in removal of the phthalimide group with concomitant lactamization to the 4-silyloxy-3-azidopiperidinones. Staudinger reduction of the azidopiperidinones followed by N-phthaloylation and hydrolysis of the silyl group afforded the 4'-hydroxylated-6'deoxythalidomide derivatives. \odot 2000 Elsevier Science Ltd. All rights reserved.

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The teratogenic and antiangiogenic properties of thalidomide (1) may be attributed to a prior metabolic activation step which produces hydroxylated derivatives of either the phthalimide or glutarimide rings.¹ The disposition of 1 and its hydroxylated analogs may also be traced through hydrolytic pathways in vivo which can operate to produce acyclic derivatives, namely the 2-Nphthalimidoglutamine, -isoglutamine, -glutamic acid and the carboxybenzamidic (phthalamic acid) derivatives of glutamine, isoglutamine and glutamic acid.2 The tendency of thalidomide to racemize under physiological conditions³ together with the sensitivity of the compound and its metabolites toward hydrolysis have complicated the task of identifying the active components and elucidating the pathway of their formation. Moreover, details regarding the molecular mechanism of thalidomide and its putative metabolites have remained elusive. Current studies in our laboratory have involved the synthesis of hydroxylated metabolites and analogs of thalidomide and its hydrolysis products⁴ with a particular emphasis on two fronts: (1) stereoselective preparation of hydroxylated products of the glutarimide ring and their analogs; and (2) preparation of deoxygenated analogs in which a carbonyl has been removed from either the glutarimide ring or the phthalimide ring thereby enhancing the biological activity of the molecule as an angiogenesis inhibitor.^{1b,5a}

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This letter details our progress in the synthesis of hydroxythalidomides 2 and 3. Analog 2 is a putative metabolic product of 4'-oxidation of the glutarimide ring, or if considering the ultimate hydrolysis products, oxidation of the β-position of glutamine, isoglutamine or glutamic acid. We designed the synthesis of 2 to include the deoxygenated analog 3 since the piperidinone analogs, as in the case of the phthalimidine analog $EM-12$ (4),⁵ may offer increased activity due to the resistance of the lactams to hydrolysis in vivo. As in the synthesis of 5'-hydroxythalidomide derivatives (5) ,⁶ a subject of previous reports from our laboratory,^{7,8} the common objective is the development and evaluation of methods for the stereoselective synthesis of β -and γ -hydroxylated a-amino acids and their derivatives.

Wittig olefination of 3-(N-phthalimido)propionaldehyde 6^9 with (benzyloxycarbonylmethyl)triphenylphosphonium bromide in the presence of NaOH/triethylamine in a two-phase system of dichloromethane/H₂O afforded the E-5-N-phthalimido- α , β -unsaturated benzyl ester 7^{10} in 99% yield (Scheme 1). Sharpless asymmetric dihydroxylation of α , β -unsaturated ester 7^{11} gave the 2R,3S-diol ester 8^{12} in 82% yield.¹³ Azido ester 10 was prepared through a displacement of the sensitive nosylate 9 by a two-step sequence which did not necessitate rigorous purification of intermediate 9. Treatment of diol ester 8 with 4-nitrobenzenesulfonyl chloride followed by rapid silica gel filtration (hexane/EtOAc) provided the sensitive α -nosyl ester 9 which was immediately exposed to sodium azide in dimethylformamide to afford the $2S,3S-N$ -phthalimido azido ester 10.^{14,15}

Scheme 1. Conditions: (a) BnOCOCH₂PPh₃⁺Br⁻, NaOH, Et₃N, CH₂Cl₂, H₂O, 12 h, rt (99%); (b) AD mix- α , $CH_3SO_2NH_2$, $(CH_3)_3COH$, H_2O , 48 h, $5^{\circ}C$ (82%); (c) $4\text{-}NO_2PhSO_2Cl$, pyr, CH_2Cl_2 , 72 h, $5^{\circ}C$; (d) NaN_3/DMF , 12 h, 60°C (59% from 8); (e) (CH₃)₃C(CH₃)₂SiOSO₂CF₃, 2,4,6-collidine, CH₂Cl₂, 12 h, 5°C (89%); (f) N₂H₄, EtOH, 12 h, rt $(90\%, 14a:14b=3:1)$; (g) PPh₃, THF, H₂O, 16 h, rt; (h) $o-C_6H_4(COCl)_2$, DMAP, CH₂Cl₂, 16 h, 0°C (53% from 10); (i) H2, Pd-C, EtOH, 7 atm, 72 h, rt (78%)

Staudinger reduction of the azido ester 10 followed by direct N-phthaloylation furnished the 2S,3S-bis-N-phthalimido- β -hydroxyester 12 (53% from 10). Submission of 12 to catalytic hydrogenation with palladium on charcoal facilitated cleavage of the benzyl group thereby affording the bis-N-phthalimido- β -hydroxyornithine derivative 13 (78%) (Scheme 2).¹⁶

Scheme 2. Conditions: (a) PPh₃, CH₃CN, 12 h, rt; then H₂O, Et₃N, 5 h, rt (68%); (b) o -C₆H₄(COCl)₂, DMAP, CH₂Cl₂, 12 h, 0° C (99%); (c) R-(+)-MTPA-Cl, DMAP, CH₂Cl₂, 24 h, 0° C to rt; (d) TFA:H₂O (9:1), CH₂Cl₂, 3 h, 0° C to rt (70%)

The alcohol function of azido ester 10 was silylated with *tert*-butyldimethylsilyl triflate¹⁷ to give the α -azido- β -silyloxy benzyl ester 11 (89%).¹⁸ Hydrazinolysis of the benzyl ester 11 resulted in smooth removal of the phthalimide group and cyclization to provide the epimeric 3-azido-4-silyloxypiperidinones 14a¹⁹ and 14b (90% in 3:1 ratio).20 Staudinger reduction/hydrolysis converted the 3S,4S-3-azido-4-silyloxypiperidinone 14a to the corresponding 3-amino-4-silyloxypiperidinone 15^{21} in 68% yield. The enantiomeric composition of the 3-amino-4-silyloxypiperidinone 15 was determined by Mosher amide analysis.²² Treatment of 15 with $R-(+)$ - α -methoxy- α -(trifluoromethyl)phenylacetyl chloride provided the chromatographically homogeneous corresponding MTPA amide 16. ¹⁹F NMR analysis of 16 revealed $>96\%$ enantiomeric purity [δ 7.13 (major); δ 7.46 (minor), 470 MHz, CDCl₃/TFA]. Phthaloylation of the 3-amino-4-silyloxypiperidinone 15 proceeded smoothly to deliver the 6'-deoxy-4'-silyloxythalidomides $17a^{23}$ and $17b$ (99%) as a mixture (17a:17b=2:1) of epimers. Desilylation of 17a and 17b under acidic conditions²⁴ provided the 6'-deoxy-4'-hydroxythalidomides $2S,3S-3a^{25}$ and $2R,3S-3b$ in 70% yield. The oxidation of the 4'-silyloxy-6'deoxythalidomides $17a$ and $17b$ at C-6' to afford the corresponding glutarimides is currently under examination. The employment of peroxide/metal oxidants such as $tert$ -butylhydroperoxide and Mn ^{III}acac systems, although fruitful in many lactam to imide oxidations,²⁶ have been largely unsuccessful in our application. Recently, Takeuchi²⁷ has reported the oxidation of 3'-fluoro-6'-deoxythalidomide analog to the corresponding glutarimide using a ruthenium dioxide/metaperiodate system. These new results have given us impetus and encouragement in our pursuit of a suitable oxidation system for the silyloxylactam substrates. In summary we have detailed a novel route to β -hydroxyl-substituted 6'-deoxythalidomide analogs via the lactamization of a pivotal hydroxyornithine derivative. Prior to lactamization to the cycloornithine derivatives, a key functionalization was introduced using the Sharpless asymmetric dihydroxylation method.

References

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- 10. For 7: IR (KBr) v_{max} 3475, 3083, 2921, 2251, 1773, 1687, 1426 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz) δ 7.82–7.81 (m, 2H), 7.34±7.29 (m, 5H), 6.95 (dt, 1H, J=15.7, 7.0 Hz), 5.92 (dt, 1H, J=15.7, 0.7 Hz), 5.14 (s, 2H), 3.81 (t, 2H, J=7.0 Hz), 2.60 (dd, 2H, J=7.0, 7.0 Hz); 13C NMR (CDCl3, 125 MHz) 168.1, 166.1, 144.8, 136.0 134.1, 132.0, 128.6, 128.2, 123.5, 123.4, 109.2, 66.2, 36.3, 31.3. HRMS (EI): calcd for $C_{20}H_{17}O_4$ [M⁺]: 335.1157; found: 335.1170.
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- 12. For 8: $[\alpha]_D^{25}$ -29.2 (c=1.08, CHCl₃); IR (KBr) v_{max} 3501, 3317, 2940, 2359, 2111, 1715, 1397 cm⁻¹; ¹H NMR (CDCl3, 500 MHz) 7.84±7.82 (m, 2H), 7.72±7.70 (m, 2H), 7.32±7.24 (m, 5H), 5.24±5.19 (m, 2H), 4.12±4.07 (m, 1H), 3.95±3.88 (m, 2H), 3.86±3.78 (m, 1H), 3.04 (d, 1H, J=11.2 Hz), 2.88 (d, 1H, J=7.4 Hz), 2.4 (m, 1H), 1.88± 1.85 (m, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 172.8, 168.7, 134.1, 132.0, 128.6, 128.5, 128.4, 128.2, 123.4, 73.6, 69.7, 67.6, 34.6, 32.5. HRMS (EI): calcd for $C_{20}H_{19}O_6$ [M⁺]: 369.1212; found: 369.1216.
- 13. Based on the Sharpless model for E-olefin enantioselectivity with AD mix- α , the configuration of diol 8 is assumed to be $2R,3S$; however, low ee's have been reported in cases of ε -substituted-E-2-pentenoate esters (Ref. 11a) and Eolefins bearing carbamate substituents (Shirota, O.; Nakanishi, K.; Berova, N. Tetrahedron 1999, 55, 13643-13658). For purposes of relating absolute configuration, the isopropylidene derivative of $\bf8$ was prepared from the O-benzylbenzoate derivative of known $(2R,3S)$ configuration: Maier, M. E.; Hermann, C. Tetrahedron 2000, 56, 557 $-$ 561 (shown below) and the configuration of 2R,3S-8 was confirmed. The enantioselectivity of the AD reaction as applied to e-substituted-E-2-pentenoate benzyl esters has been under examination in these laboratories (Ref. 8a). These results will be reported in a separate communication with full experimental details.

- 14. Attempted derivatization of the β -hydroxyl function of 10, for assay of chiral purity, with $R-(+)$ -MTPA chloride (DMAP/CH₂Cl₂/0^oC) resulted in facile elimination to the corresponding α , β -unsaturated ester (β -H, t, 6.23 δ , 500 MHz, CDCl₃). See Ref. 22 for the use of $R-(+)$ -MTPA chloride.
- 15. For 10: $\left[\alpha\right]_D^{25}$ -45.11 (c=1.74, CHCl₃); IR (Kbr) v_{max} 3466, 3349, 2114, 1769, 1703, 1404, 1199 cm⁻¹; ¹H NMR (CDCl3, 500 MHz) 7.84±7.82 (m, 2H), 7.72±7.70 (m, 2H), 7.32±7.24 (m, 5H), 5.24±5.17 (m, 2H), 4.11±4.09 (m, 1H), $4.02-4.01$ (d, 1H, $J=5.5$ Hz), $3.93-3.91$ (br m, 1H), $3.82-3.82$ (m, 2H), $1.86-1.83$ (m, 2H); ¹³C NMR (CDCl₃,

125 MHz) 172.8, 168.7, 134.1, 132.0, 128.6, 128.5, 128.4, 128.2, 123.4, 73.6, 69.7, 67.6, 34.6, 32.5. MS: [ES+] m/z 395.3 ([M+1], 100%).

- 16. Bis-phthalimide 13 was inactive when evaluated for inhibition of angiogenesis in the rat aortic ring and human aortic endothelial cell (HAEC) bioassays (see Ref. 1b).
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- 18. For 11: $[\alpha]_D^{25}$ +8.28 (c = 1.98, CHCl₃); IR (KBr) v_{max} 3033, 2945, 2858, 2112, 1715, 1443, 1398, 775 cm⁻¹; ¹H NMR $(CDCl₃, 500 MHz)$ δ 7.82-7.80 (m, 2H), 7.70-7.67 (m, 2H), 7.33-7.28 (m, 5H), 5.28-5.11 (m, 2H), 4.19-4.14 (m, 1H), 3.81±3.75 (m, 1H), 3.73±3.66 (m, 1H), 2.08±1.99 (m, 1H), 1.91±1.84 (m, 1H), 0.88 (s, 9H), 0.12 (s, 3H), 0.08 (s, 3H); 13C NMR (CDCl3, 125 MHz) 167.8, 135.2, 134.2, 131.8, 128.8, 123.5, 72.2, 71.3, 67.9, 66.6, 64.6, 34.6, 32.0, 26.0, 18.0. MS: [ES+] m/z 509.4 ([M+1], 100%).
- 19. For 14a: $[\alpha]_D^{25}$ -24.06 (c=1.65, CHCl₃); IR (KBr) v_{max} 3350, 3311, 3199, 2929, 2247, 2105, 1680, 1255 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz) 6.65 (br s, 1H), 4.27 (br m, 1H), 3.72 (d, 1H, J=2.8 Hz), 3.59-3.63 (m, 1H), 3.23-3.19 $(m, 1H)$, 1.96 $(m, 1H)$, 1.90 $(m, 1H)$, 0.88 (s, 9H), 0.14 (s, 3H), 0.09 (s, 3H); ¹³C NMR (CDCl₃, 125 MHz) δ 168.7, 68.4, 63.1, 37.7, 28.6, 25.7, 16.5, ^5.0.
- 20. The hydrazinolysis of 11 to $14a/14b$ is the most facile conversion of an ornithine derivative to the corresponding d-valerolactam that we have observed. For example, see: Toshima, H.; Maru, K.; Saito, M.; Ichiara, A. Tetrahedron 1999, 55, 593–5808. Alternatively, the N-phthaloyl-3-azidoalcohol 10 could be lactamized with hydrazine to the corresponding 4-hydroxy-3-azidopiperidinone 18 (below); however, the yields of the cyclization and subsequent silylation of the hydroxyl group to afford 14a and 14b were lower. In addition, a remarkable tendency for the 4-hydroxy-3-azidopiperidinone to undergo elimination to the 2-azido-2,3-didehydrocycloornithine 19 (Yonezowa, Y.; Hirosaki, T.; Hayashi, T.; Shin, C. Synthesis 2000, 1, 144-148) was observed when attempting thionoformylation (below) of the alcohol function (phenylthionochloroformate, DMAP, CH₂Cl₂, 0° C) en route to a $N-R-(+)$ -MPTA-cycloornithine amide derivative 20, by a Barton deoxygenation/reduction/MTPA amide derivatization sequence (shown retrosynthetically).

- 21. For 15: $\left[\alpha\right]_D^{25}$ +12.58 (c=1.2, CHCl₃); IR (KBr) v_{max} 3246, 3202, 2973, 2960, 1672, 1488, 1255 cm⁻¹; ¹H NMR $(CDCl₃, 500 MHz)$ δ 6.65 (br s, 1H), 4.27-4.26 (br m, 1H), 3.72 (d, 1H, J=2.8 Hz), 3.56 (dt, 1H, J=4.9, 6.2, 16.0 Hz), 3.23–3.19 (m, 1H), 1.96–1.92 (m, 1H), 1.90–1.84 (m, 1H), 0.88 (s, 9H), 0.14 (s, 3H), 0.09 (s, 3H); ¹³C NMR $(CDCl₃, 125 MHz)$ δ 168.7, 68.4, 63.2, 37.7, 28.7, 25.8, 18.1, -5.0. MS: [ES+] m/z 245.3 ([M+1], 100%).
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- 23. For 17a: $[\alpha]_D^{25}$ +39.0 (c=1.1, CHCl₃); IR (KBr) ν_{max} 3219, 3098, 2956, 2860, 1772, 1715, 1687, 1393, 767 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz) δ 7.80 (br s, 2H), 7.67 (dd, 2H, J=3.0, 5.3 Hz), 6.79 (br s, 1H), 4.75 (d, 1H, J=4.5 Hz), 4.31–4.29 (m, 1H), 3.79–3.73 (m, 1H), 3.25–3.20 (m, 1H), 2.0–1.98 (m, 2H), 0.72 (s, 9H), -0.08 (s, 3H), -0.31 (s, 3H); ¹³C NMR (CDCl₃, 125 MHz) δ 168.0, 134.1, 132.0, 123.4, 67.7, 57.4, 38.3, 32.0, 25.4, 17.6, -4.7, -5.2. MS: m/z 375 [M]⁺ (82%), 145 (100%).
- 24. While tetra-N-butylammonium fluoride (TBAF) facilitated cleavage of the tert-butyldimethylsilyl (TBDMS) group, the quaternary ammonium by-products complicated the chromatographic purification of the 6'deoxythalidomides 3a/3b thereby necessitating the employment of the easily-removable TFA.
- 25. For 3a: $[\alpha]_D^{25}$ +16.6 (c = 1.0, MeOH); IR (KBr) v_{max} 3346, 2930, 2860, 1713, 1666, 1399, 1082, 717 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz) δ 7.87 (br d, 2H, J=3.0 Hz), 7.83 (br d, J=2.2 Hz), 4.90 (d, 1H, J=4.5 Hz), 4.31 (m, 1H), 3.76-3.70 (m, 1H), 3.32-3.29 (m, 1H), 2.12-2.11 (m, 2H); ¹³C NMR (CDCl₃, 125 MHz) δ 170.2, 169.5, 136.1, 135.5, 133.4, 124.3, 67.7, 58.5, 39.4, 31.9. MS: [ES+] m/z 261.3 ([M+1], 100%).
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